

OPTIMIZATION AND KINETICS STUDIES OF THE DISSOLUTION OF DOLOMITE IN SULPHURIC ACID (H₂SO₄) VIA BOX-BEHNKEN EXPERIMENTAL DESIGN

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ERRATUM

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Where it reads: Dolomite is a naturally occurring double carbonate mineral of magnesium and calcium with the chemical formula $MgCa(CO_3)_2$ (Baba et al., 2014; Pultar et al., 2018; Ajala et al., 2019). Dolomite ore can be used directly as an accelerating agent for cement hydration in concrete and a neutralizing agent in acidic soil (Mubaroka and Kurniawanb, 2015; Ajala et al., 2019). In addition, dolomite is rich in magnesium. Hence it can be used for the production of magnesium-based compounds such as magnesite ($MgCO_3$), magnesium oxide (MgO), magnesium sulfate, as well as calcium sulfate, commonly known as gypsum ($CaSO_4$). The abundant dolomite of Nigeria is spread across Abuja and the following states: Kogi, Oyo, Yobe, Kwara, Edo, and Nasarawa (Mookah and Abolarin, 2005).

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The electronic copy of the manuscript was update and this modification was included in the file.